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(i) Tetraammineaquacobalt(III) chloride

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NCERT TEXTBOOK QUESTIONS SOLVED. 3.1. How would you determine the standard electrode potential of the system $\text{Mg}^{2+}|\text{Mg}$? Ans: A cell will be set up consisting of Mg/MgSO_4 (1 M) as one electrode and standard hydrogen electrode Pt, H_2 (1 atm) $|\text{H}^+$ (1 M) as second electrode, measure the EMF of the cell and also note the direction of deflection in the voltmeter.

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12 Chemistry. Solid State NCERT InText Solution. Question: 1.1 - Why are solids rigid? Answer: The particles of solids are close packed and can only oscillate about their fixed positions. These properties make solids rigid.

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