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 There are three states of matter that we will learn about in this chapter. (If you want to learn about more states of matter, I can refer you to somebody.) Those three states are solid, liquid, and gas. These three states are quite different. The main difference is in their particles.

Chapter 13: States of Matter - Chemistry by Anna
 Section 13.4 Changes of State The relationship among the solid, liquid, and vapor states (or phases) of a substance in a sealed container are best represented in a single graph called a phase diagram Phase diagram- gives the temperature and pressure at which a substances exists as solid, liquid, or gas (vapor) 70.

Chemistry - Chp 13 - States of Matter
 Vocabulary from Chapter 13 "Phases of Matter" for Mr. Testa's Honors Chemistry class. Still a work in progress. Currently only contains some gas laws and properties.

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 Chapter 13 States of Matter137 SECTION 13.1 THE NATURE OF GASES (pages 385 – 389) This section introduces the kinetic theory and describes how it applies to gases. It defines gas pressure and explains how temperature is related to the kinetic energy of the particles of a substance. Kinetic Theory and a Model for Gases (pages 385 – 386) 1.

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