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For most chemical reactions in

this book, solids are labeled (s),

liquids (l), and gases (g). The

numerical coefficients in front of

the chemical formulas express

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Study moles of each compound or element. The preceding reaction can be interpreted in terms of moles or masses. (See Table 1.)

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11.1 Describing Chemical Reactions 11

During chemical reactions, the molecules of a substance are

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broken apart and rearranged into new and different molecules. It would be like taking apart the hydrogen and oxygen atoms of water, and...

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(pages 25–28) 13. The space in
which the electrons move is huge
compared to the space occupied
by the . 14. The electrons farthest
from the nucleus or most loosely
held are called . 15.

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Conservation of Mass ERQ Target
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chemical reaction is releasing or
absorbing energy (thermal).
Exothermic & Endothermic
Reactions Notes Yeast & Peroxide
Lab Hot & Cold Packs Lab Target
5: I can design, test and modify a
device that involves a chemical
reaction that releases or absorbs

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Classifying Chemical Reactions (pp. 200-201) 21. In what three categories can chemical reactions be classified? 22.

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It might indicate a chemical reaction, because the bubbles are gas, and the presence of a gas indicates a new substance may have been formed. 12. True 13. A change in temperature reflects a change in energy. 14. False 15. a. Energy is absorbed b. Exothermic

6.2 Describing Chemical Reactions Guided Reading and Study Use Target Reading Skills a

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