

Chapter 8 Covalent Bonding Workbook Answers

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Chapter 8 Covalent Bonding 183 Section Review Objectives • State a rule that usually tells how many electrons are shared to form a covalent bond • Describe how electron dot formulas are used • Predict when two atoms are likely to be joined by a double or a triple covalent bond • Distinguish between a single covalent bond and other covalent bonds

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Chapter 8 Covalent Bonding. chemical bond. covalent bond. molecule. electron dot diagram. ???the force that holds two atoms together. ???the chemical bond that results from sharing valence electro.... ??a molecule is formed when two or more atoms bond covalently. ???used to show valence electrons of atoms.

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